# Organic & Biomolecular Chemistry



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Cite this: DOI: 10.1039/d1ob01927a

# Perforalactones D and E, two new C-20 quassinoids with potential activity to induce lysosomal biogenesis from the twigs of *Harrisonia perforata* (Blanco) Merr.<sup>†</sup>

Received 29th September 2021, Accepted 20th October 2021 DOI: 10.1039/d1ob01927a

rsc.li/obc

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Two new quassinoids (1 and 2) were isolated from the twigs of *Harrisonia perforata* (Blanco) Merr. Perforalactone E (2) possesses an uncommon hexacyclic  $1\alpha$ , $12\alpha$ : $5\alpha$ , $13\alpha$ -dicyclo- $9\beta$ H-picrasane skeleton. Its structure was determined based on spectroscopic data and X-ray crystallography. Compounds 1 and 2 could significantly induce lysosomal biogenesis through transcriptional activation of lysosomal genes.

Quassinoids, originating from the oxidative degradation of tetracyclic tirucallane triterpene, are a diverse class of secondary metabolites identified from nature and mostly belong to the Simaroubaceae family.<sup>1</sup> They have attracted wide-spread attention due to their both biological and pharmacological effects including antitumor, antimalarial, antiviral, antiinflammatory, insecticidal, amoebicidal, anthelmintic, and antiulcer activities.<sup>1-6</sup> Previous studies on the chemical components of plants from the species *Harrisonia perforata* (Blanco) Merr. have resulted in a large array of quassinoids, limonoids, chromones and polyketides.<sup>7-16</sup>

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<sup>e</sup>Guizhou Chemical Drug Research and Development Engineering Technical Center, Guizhou Medicinal University, Guiyang 550004, China Recently, we found that the analogues of 20-deoxyingenol could induce lysosomal biogenesis and clear  $\beta$ -amyloids in the brain of mice, suggesting that these compounds have the potential to be developed for the treatment of neurodegenerative diseases.<sup>17</sup> To identify more natural products which could induce lysosomal biogenesis, we studied the chemical composition of the twigs of *Harrisonia perforata*. Two undescribed natural products, perforalactones D (1)§ and E (2)¶, were obtained. In addition, the lysosomal biogenesis activities of compounds 1 and 2 were evaluated. In this paper, we describe the isolation, structural determination and biological evaluation of these two compounds (Fig. 1).

Compound 1 was obtained as white powder and has the molecular formula  $C_{20}H_{24}O_5$  as determined by HRESIMS (*m/z* 367.1518 [M + Na]<sup>+</sup>, (calcd for  $C_{20}H_{24}O_5$ Na, 357.1516)) with nine degrees of unsaturation. The IR spectrum suggested the presence of hydroxyl (3435 cm<sup>-1</sup>) and  $\alpha$ , $\beta$ -unsaturated carbonyl (1729 cm<sup>-1</sup>) functionalities. In its <sup>1</sup>H NMR spectrum, **1** showed signals due to the protons of four quaternary methyl groups [ $\delta_H$  1.20 (3H, s, Me-19), 1.24 (3H, s, Me-20), 1.96 (3H, s, Me-18), and 1.97 (3H, s, Me-21)] and two olefinic protons [ $\delta_H$  6.07 (1H, dd, *J* = 2.9, 1.4 Hz, H-3) and 5.89 (1H, dd, *J* = 1.5 Hz, H-12)] (Table 1). The <sup>13</sup>C NMR spectrum of **1** (Table 1), analyzed with the help of DEPT and HSQC data, showed 20 signals assignable to four methyls ( $\delta_C$  10.7, 21.8, 22.4, and 22.8), two methylenes ( $\delta_C$  24.9 and 30.8), seven methines ( $\delta_C$ 



Fig. 1 Chemical structures of perforalactones D (1) and E (2).

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<sup>&</sup>lt;sup>†</sup>Electronic supplementary information (ESI) available: Experimental procedures and NMR spectra. CCDC 2110146. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/d1ob01927a

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No.	Perforalactone D (1)		Perforalactone E (2)	
	$\delta_{ m C}$	$\delta_{ m H}$	$\delta_{ m C}$	$\delta_{ m H}$
1	85.5 CH	3.97 (d, 3.8)	85.4 C	
2	197.2 C		195.0 C	
3	126.7 CH	6.07 (dd, 2.9,1.4)	126.3 CH	6.38 (d, 1.1)
4	160.1 C		166.5 C	
5	42.6 CH	2.95, s	49.9 C	
6	24.9 CH <sub>2</sub>	(a) 2.11 (dd, 9.2, 6.7) (b) 2.09 (d, 2.8)	29.9 CH <sub>2</sub>	(a) 2.00 (dd, 15.7, 5.3) (b) 2.57 (dd, 15.7, 10.2)
7	82.4 CH	4.44 (dd, 3.9,2.2)	75.7 CH	4.36 (ddd, 10.2, 5.3,2.8
8	38.2 C		39.6 C	
9	51.7 CH	2.98, s	57.0 CH	1.90 (d, 1.8)
10	46.2 C		60.1 C	
11	199.3 C		206.4 C	
12	127.2 CH	5.89 (dd, 1.5)	67.9 CH	2.17 (d, 1.8)
13	157.2 C		53.0 C	
14	47.8 CH	2.42 (dd, 12.5, 6.7)	39.4 CH	2.42 (dt, 10.4, 2.8)
15	30.8 CH <sub>2</sub>	(a) 3.02 (dd, 18.7, 6.7) (b) 2.59 (dd, 18.7, 12.5)	28.9 CH <sub>2</sub>	(a) 2.77 (dd, 19.7, 10.4) (b) 2.66 (dd, 19.7, 2.8)
16	168.3 C		169.7 C	
18	22.8 CH <sub>3</sub>	1.96, s	23.0 CH <sub>3</sub>	2.05 (d, 1.1)
19	10.7 CH <sub>3</sub>	1.20, s	12.1 CH <sub>3</sub>	0.95, s
20	22.4 CH <sub>3</sub>	1.24, s	19.6 CH <sub>3</sub>	1.19, s
21	21.8 CH <sub>3</sub>	1.97, s	18.7 CH <sub>3</sub>	1.02, s
OH		5.05 (d, 3.9)		4.25, s

Table 1 <sup>1</sup>H (500 MHz) and <sup>13</sup>C (150 MHz) NMR data for 1 and 2 in CDCl<sub>3</sub>

42.6, 47.8, 51.7, 82.4, 85.5, 126.7, and 127.2), and 7 nonprotonated carbons including two ketone carbonyls ( $\delta_{\rm C}$  197.2 and 199.3), one lactone carbonyl ( $\delta_{\rm C}$  168.3), two olefinic carbons ( $\delta_{\rm C}$  126.7 and 127.2), and four quaternary carbons ( $\delta_{\rm C}$  38.2, 46.2, 157.2, and 160.1) characteristic of tetracyclic C20-type quassinoids. The <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data of **1** resembled those of 11-dehydroklaineanone,<sup>18–20</sup> and suggested that an additional trisubstituted double bond in **1** replaced two methines and an oxygenated methine in the latter. The  $\Delta^{12,13}$  double bond was confirmed by HMBC correlations from H-12 to C-9 and C-14, and from H<sub>3</sub>-21 to C-12, C-13, and C-14 (Fig. 2). In addition, the relative configuration at the remaining



Fig. 2 Key  ${}^{1}H - {}^{1}H$  COSY, HMBC, and ROESY correlations of compounds 1 and 2.

chiral centers of 1 would be analogous to those of the latter on the basis of  $^{13}$ C NMR shifts and ROE data (Fig. 2).

Compound 2 was obtained as a colourless crystal, and its molecular formula was determined to be C<sub>20</sub>H<sub>22</sub>O<sub>5</sub> by HRESIMS at m/z 341.1399  $[M - H]^-$  (calcd 341.1394), with ten degrees of unsaturation. The IR spectrum suggested the presence of hydroxyl (3429 cm<sup>-1</sup>),  $\alpha$ , $\beta$ -unsaturated carbonyl (1737 cm<sup>-1</sup>) and  $\delta$ -lactone (1766 cm<sup>-1</sup>). The <sup>1</sup>H NMR spectrum of 2 (Table 1) displayed signals for four methyl groups at  $[\delta_{\rm H}]$ 0.95 (3H, s, Me-19), 1.02 (3H, s, Me-21), 1.19 (3H, s, Me-20), and 2.05 (3H, d, J = 1.1 Hz, Me-18)] and one olefinic proton [ $\delta_{\rm H}$ 6.38 (1H, d, J = 1.1 Hz, H-3)]. The <sup>13</sup>C- and DEPT NMR spectra showed twenty carbon signals due to four methyls ( $\delta_{\rm C}$  12.1, 18.7, 19.6, and 23.0), two methylenes ( $\delta_{\rm C}$  28.9 and 29.9), five methines ( $\delta_{\rm C}$  39.4, 57.0, 67.9, 75.7, and 126.3) and nine nonprotonated carbons representing two ketone carbonyls ( $\delta_{\rm C}$ 195.0 and 206.4), one ester/lactone carbonyl ( $\delta_{\rm C}$  169.7), one olefinic carbon ( $\delta_{\rm C}$  166.5), one oxygenated carbon ( $\delta_{\rm C}$  85.4), and four quaternary carbons ( $\delta_{\rm C}$  39.6, 49.9, 53.0, and 60.1). Apart from four degrees of unsaturation occupied by one double bond and three carbonyls, the remaining six were assumed for the presence of the hexacyclic system of 2.

Compound 2 was deduced to feature a rare 6/5/6/5/6/6 hexacyclic ring system by detailed analysis of its 1D and 2D NMR spectra. The <sup>1</sup>H–<sup>1</sup>H COSY and HSQC spectra of 2 revealed the existence of two spin systems (C-6–C-7 and C-14–C-15) as shown in bold in Fig. 2. According to the key HMBC correlations from OH-1 to C-1/C-2/C-10, from H-3 to C-1, from H<sub>2</sub>-6 to C-4/C-5, from H-7 to C-16, from H<sub>3</sub>-20 to C-7/C-8/C-9, from H-9 to C-11/C-12, from H<sub>3</sub>-19 to C-1/C-9/C-10, and from H<sub>3</sub>-21 to C-12/C-13/C-14, a highly functionalized C20-type quassinoid structure, which is the same as that of compound 1, was estab-



Fig. 3 Structure of 2 from X-ray crystallographic analysis.

lished as shown in Fig. 2. The construction of 2 left four unoccupied conjugating sites, including C-1, C-5, C-12, and C-13. The C-1/C-12 and C-5/C-13 linkages in 2 were determined by HMBC correlations from H<sub>3</sub>-21 to C-5 and from H-12 to C-10. Thus, 2 was determined to have an unusual hexacyclic  $1\alpha$ ,12 $\alpha$ :5 $\alpha$ ,13 $\alpha$ -dicyclo-9 $\beta$ H-picrasane skeleton. To the best of our knowledge, only two compounds, shinjulactone C and ailantinol E, with the same skeleton were isolated from the root bark of *Ailanthus altissima*.<sup>21,22</sup> The structure of 2, named perforalactone E, was finally confirmed by single-crystal X-ray diffraction, which also unambiguously determined the absolute configuration to be 1*S*,5*S*,7*R*,8*S*,9*R*,10*R*,12*S*,13*R*,14*S* (Fig. 3) with a Flack parameter of 0.10(3).

Structure comparisons of **1** and **2** suggested a possible route for the biosynthesis of **2** from **1**. Herein, we propose a putative biogenetic pathway for **2**. **1** undergoes inversion of the chiral center at the C-9 position and then catalyzed by enzymes to produce a pentadienyl cation. Compound **2** can then be produced by intramolecular ionic cycloaddition between the pentadienyl cation and an olefin as shown in Fig. 4.

The new compounds 1 and 2 were assessed for their activity to induce lysosomal biogenesis through LysoTracker Red staining. We used Hep14, that could induce lysosomal biogenesis in our previous study,<sup>17</sup> and rapamycin as positive controls. The U251 cells were treated with compounds 1 and 2 for 24 h at 10  $\mu$ M, and compounds 1 and 2 increased the LysoTracker staining intensity. The induction of lysosomal biogenesis was increased in a concentration-dependent manner, with the greatest increase observed at 40  $\mu$ M (Fig. 5A and B). The transcriptional levels of lysosome-related genes including the lyso-



Fig. 4 Proposed biogenetic relationship of compounds 1 and 2.



Fig. 5 Compounds 1 and 2 affect lysosomal biogenesis. (A) Representative images of U251 cells stained with LysoTracker Red that were treated with compounds 1 and 2 at the indicated concentrations. Scale bars represented 5  $\mu$ m in these images. (B) Quantification of lysosomes induced by compounds 1 and 2. Compounds 1 and 2 induced lysosomes in a concentration-dependent manner (fold induction of LysoTracker Red staining). Hep14 and rapamycin were used as positive controls. Three independent experiments were repeated with similar results. (C) Compounds 1 and 2 induced the mRNA levels of lysosomal genes. The bar graph shows the relative mRNA level of each gene, and GAPDH serves as the internal control. Student's t-test was used to check the statistical difference (relative to the control [untreated cells]). *P* < 0.05 was considered statistically significant.

somal membrane protein (LAMP1), LAMP2, lysosomal hydrolytic enzymes (CTSB, CTSD), lysosomal ATPase  $H^+$  transporting V0 subunit E1 (ATP6V0E1) and lysosomal sulfatase (ARSB) were also checked. As shown in Fig. 5C, the mRNA levels of these genes were upregulated at 24 h after the cells were treated with compounds 1 and 2, respectively. These data further demonstrated that compounds 1 and 2 can induce lysosomal biogenesis.

### Conclusions

In this study, two new quassinoids (1 and 2) were isolated from the twigs of *H. perforata*. 2 possesses a special cage-like carbon skeleton. The structure and absolute configuration of 2 were confirmed by single-crystal X-ray diffraction. To the best of our knowledge, this is the first time that this skeleton type has been isolated from *H. perforata* and it is the third quassinoid with a cage-like skeleton. Moreover, compounds 1 and 2 could significantly induce lysosomal biogenesis through transcriptional activation of lysosomal genes.

# Conflicts of interest

There are no conflicts of interest to declare.

# Acknowledgements

This work was financially supported by the National Natural Science Foundation of China (No. 31770392 to Y.T.D. and 31900737 to R.C.L.), the Original Innovation Project "from 0 to 1" of the Basic Frontier Scientific Research Program, CAS (ZDBS-LY-SM031 to R.C.L.), the Science and Technology Program of Yunnan Province (2018ZF013 to Y.T.D.), the CAS "Light of West China" Program (2020000023 to R.C.L.), and the Youth Innovation Promotion Association of CAS (2021000011 to R.C.L.).

# Notes and references

§ Perforalactone D (1): white powder;  $[α]_D^{21} = +35.4$  (*c* 0.07, MeOH); UV (MeOH)  $λ_{max}$  (log ε) 238 (6.06) nm; IR (KBr)  $ν_{max}$  3435, 2958, 2919, 2851, 1729, 1675, 1436, 1329, 1258, 1234, 1192, 1148, 1130, 1043, 949, 905, 803 cm<sup>-1</sup>; CD (MeOH)  $λ_{max}$  (Δε): 214 (2.03), 231 (0.58), 248 (4.77) nm; for <sup>1</sup>H and <sup>13</sup>C NMR data see Table 1; HRESIMS (positive) *m*/*z* 367.1518 [M + Na]<sup>+</sup> (calcd for C<sub>20</sub>H<sub>24</sub>O<sub>5</sub>Na, 367.1516).

 $\label{eq:performation} \begin{array}{l} \mbox{Perforalactone E (2): a colorless crystal; } [\alpha]_{20}^{D} = -220.3 \ (c \ 0.10, \ CHCl_3); \ UV \ (MeOH) \ \lambda_{max} \ (\log \varepsilon) \ 247 \ (6.38) \ nm; \ IR \ (KBr) \ \nu_{max} \ 3429, \ 3064, \ 2967, \ 2934, \ 2918, \ 2879, \ 1766, \ 1737, \ 1651, \ 1453, \ 1436, \ 1383, \ 1181, \ 1068, \ 1027, \ 1018, \ 1009, \ 979 \ cm^{-1}; \ CD \ (MeOH) \ \lambda_{max} \ (\Delta\varepsilon): \ 248 \ (9.21), \ 288 \ (-12.31) \ nm; \ for \ ^1H \ and \ ^{13}C \ NMR \ data \ see \ Table \ 1; \ HRESIMS \ (negative) \ m/z \ 341.1399 \ [M \ -H]^- \ (calcd \ for \ C_{20}H_{21}O_5, \ 341.1394). \end{array}$ 

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# **Supporting information**

# Perforalactones D and E, two new C-20 quassinoids with potential

# activity in inducing lysosomal biogenesis from twigs of Harrisonia

# perforata (Blanco) Merr.

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#### **General experimental procedures**

CD spectra were determined on the Applied Photophysics circular dichroism spectrometer (Applied Photophysics, Leatherhead, Surrey, UK). Optical rotation measurements were conducted with a Jasco P-1020 automatic polarimeter. IR spectra were surveyed on a Bio-Rad FTS-135 as KBr pellets. <sup>1</sup>H, <sup>13</sup>C, <sup>1</sup>H-<sup>1</sup>H COSY, HSQC, HMBC and ROESY spectra were collected on Bruker DRX-500 instruments (Bruker, Bremerhaven, Germany). Semi-preparative HPLC separations were performed on an Agilent 1260 liquid chromatograph (Agilent Technologies, USA) with a Waters XSelect CSH C-18 column (5  $\mu$ m, 10×250 mm). Analytical TLC systems were carried out on silica gel 60 F254 plates (Qingdao Marine Chemical Inc., Qingdao, China). Column chromatography (CC) was performed using silica gel (200-300 mesh and 60-80 mesh, Qingdao Marine Chemical, Inc., Qingdao, China) and Sephadex LH-20 (40-70  $\mu$ m, Amersham Pharmacia Biotech AB, Uppsala, Sweden). Lichroprep RP-18 gel (40-63  $\mu$ m; Merck, Darmstadt, Germany). And spots were visualized by heating silica gel plates sprayed with 5% H<sub>2</sub>SO<sub>4</sub> in EtOH.

#### **Plant material**

The branches of *Harrisonia perforata* (Blanco) Merr. were collected from Hainan Province, China, in January 2018. The plant samples were identified by Prof. Sheng-Zhuo Huang from Institute of Tropical Biotechnology, Chinese Academy of Tropical Agricultural Science. A voucher specimen (NO. 20180104) was deposited at the State Key Laboratory of Phytochemistry and Plant Resource in West China, Kunming Institute of Botany, Chinese Academy of Science (CAS).

#### **Extraction and Isolation.**

The air-dried powder of the plant material (100 kg) was extracted with 95% EtOH under reflux thrice. The crude extract was obtained by reflux. After suspension in water,

the combined extract was successively partitioned with petroleum ether, and ethyl acetate. The ethyl acetate extract (980 g) was then subjected to MCI gel column eluted with MeOH-H<sub>2</sub>O (3:7 to 10:0) to give five major fractions (Fr1–Fr5). Fr3 (45.6g) was then chromatographed on a silica gel column eluted with PE-EtOAc (from 1:0 to 1:1), to give five subfractions (Fr3-1- Fr3-5). The fraction Fr3-2 was subjected to a C18 silica gel column (MeOH/H<sub>2</sub>O 3:7 to 10: 0) and further purified by Sephadex LH-20 (MeOH) and semipreparative HPLC (CH<sub>3</sub>CN/H<sub>2</sub>O, 32:68) to obtain compounds **1** (6.1 mg) and **2** (15.6 mg).

#### X-ray crystallographic data

Single crystal culture and confirmation: First, compound 2 was added to a bottle and dissolved by the addition of MeOH/H<sub>2</sub>O (10:1). Then, the bottle was sealed with parafilm, which only reserves two tiny holes on it, then remained at room temperature for 3 days. Some crystals appeared, and for single crystal parsing, crystals were selected with sizes of 0.460m x 0.370m x 0.360m. All diffraction data were obtained on a Bruker Smart Apex CCD diffractometer equipped with graphite-monochromated Cu K $\alpha$  radiation. CCDC-2110146 (2), contain the supplementary crystallographic data. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre (http://www.ccdc.cam.ac.uk/). Thermal ellipsoids are shown at the 30% level.



View of the pack drawing of **2**. Hydrogen-bonds are shown as dashed lines.



View of a molecule of **2** with the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

Crystal data for **2** (C<sub>20</sub>H<sub>22</sub>O<sub>5</sub>)•3(H<sub>2</sub>O), M = 738.80, a = 10.2285(3) Å, b = 10.2285(3)Å, c = 31.7311(10) Å,  $a = 90^{\circ}$ ,  $\beta = 90^{\circ}$ ,  $\gamma = 90^{\circ}$ , V = 3319.8(2) Å<sup>3</sup>, T = 101.(2) K, space group *P*41212, Z = 4,  $\mu$ (Cu K $\alpha$ ) = 0.913 mm<sup>-1</sup>, 29917 reflections measured, 3278 independent reflections ( $R_{int} = 0.0226$ ). The final  $R_I$  values were 0.0575 ( $I > 2\sigma(I)$ ). The final  $wR(F^2)$  values were 0.1702 ( $I > 2\sigma(I)$ ). The final  $R_I$  values were 0.0575 (all data). The final  $wR(F^2)$  values were 0.1702 (all data). The goodness of fit on  $F^2$  was 1.120. Flack parameter = 0.10(3).

Identification code	global		
Empirical formula	$C_{40}H_{50}O_{13}$		
Formula weight	738.80		
Temperature	101(2) K		
Wavelength	1.54178 Å		
Crystal system	Tetragonal		
Space group	P4 <sub>1</sub> 2 <sub>1</sub> 2		
Unit cell dimensions	$a = 10.2285(3) \text{ Å}$ $\alpha = 90^{\circ}.$		
	$b = 10.2285(3) \text{ Å} \qquad \beta = 90^{\circ}.$		
	$c = 31.7311(10) \text{ Å}$ $\gamma = 90^{\circ}.$		
Volume	$3319.8(2) \text{ Å}^3$		
Z	4		
Density (calculated)	$1.478 \text{ Mg/m}^3$		
Absorption coefficient	0.913 mm <sup>-1</sup>		
F(000)	1576		
Crystal size	0.460 x 0.370 x 0.360 mm <sup>3</sup>		
Theta range for data collection	4.54 to 72.34°.		
Index ranges	-12<=h<=12, -12<=k<=12, -39<=l<=17		
Reflections collected	29917		
Independent reflections	3278 [R(int) = 0.0226]		
Completeness to theta = $72.34^{\circ}$	99.6 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.73 and 0.67		

## Table S1. Crystal data and structure refinement for 2

Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	3278 / 0 / 245	
Goodness-of-fit on F <sup>2</sup>	1.120	
Final R indices [I>2sigma(I)]	R1 = 0.0575, wR2 = 0.1702	
R indices (all data)	R1 = 0.0575, wR2 = 0.1702	
Absolute structure parameter	0.10(3)	
Largest diff. peak and hole	1.357 and -0.930 e.Å <sup>-3</sup>	

#### **Biological Assay**

#### Lysosome-Tracker Red staining

The U251 cells cultured in Lab-Tek II Chamber Slide were treated with compound **1** or **2** and then stained by lysosome-tracker red (500 nM) for 1 hour. For better livecell imaging, glass-bottom dishes were also used in cell growing through Olympus FV1000 confocal microscope. Images were analyzed with FV10-ASW 2.1 Viewer.

#### Quantitative real-time PCR

Total RNA was isolated from the U251 cells treated with or without compounds **1** and **2** using TRIZOL (Invitrogen, 15596-018). Around 1.5  $\mu$ g total RNA was used to synthesize single-strand cDNA using the M-MLV Reverse Transcriptase (Promega, M170A) in a final volume of 25  $\mu$ L according to the manufacturer's instructions. The relative mRNA levels of *LAMP1*, *LAMP2* and, *CTSB*, *ATP6V0E1*, *ARSB*, and *CTSD* were quantified by using quantitative real-time PCR (qRT-PCR), with normalization to the *GAPDH* gene. The qRT-PCR was performed in a total volume of 20  $\mu$ L containing 2  $\mu$ L of diluted products, 10  $\mu$ L of SYBR Master Mix (Takara), 0.2 uL 10  $\mu$ M each primer (**Table S2**), on an BIO-RAD Real-time PCR detection system. The qRT-PCR thermal cycling conditions were composed of a denaturation cycle at 95°C for 5 min, followed by 40 cycles of 95 °C for 10 s and 58 °C for 30 sec.

## Statistics and reproducibility

Data analysis was carried out using GraphPad Prism 8. The Student's *t*-test was used to detect the mRNA expression difference between groups. Values were expressed as mean  $\pm$  standard error. \*, P < 0.05; \*\*, P < 0.01; \*\*\*, P < 0.001; \*\*\*\*, P < 0.0001; ns, not significant.

Primer	Sequence (5'-3')	Product length (bp)
LAMP1 Forward	TCTCAGTGAACTACGACACCA	151
LAMP1 Reverse	AGTGTATGTCCTCTTCCAAAAGC	
LAMP2 Forward	GAAAATGCCACTTGCCTTTATGC	184
LAMP2 Reverse	AGGAAAAGCCAGGTCCGAAC	
CTSB Forward	ACAACGTGGACATGAGCTACT	85
CTSB Reverse	TCGGTAAACATAACTCTCTGGGG	
ATP6V0E1 Forward	GTCCTAACCGGGGGAGTTATCA	101
ATP6V0E1 Reverse	AAAGAGAGGGTTGAGTTGGGC	
ARSB Forward	TCTTGCTGGCAGACGACCTA	121
ARSB Reverse	GGCTGCGTGTAGTAGTTGTCC	
CTSD Forward	CACCACAAGTACAACAGCGAC	77
CTSD Reverse	CCCGAGCCATAGTGGATGT	
GAPDH Forward	CTGGGCTACACTGAGCACC	101
GAPDH Reverse	AAGTGGTCGTTGAGGGCAATG	

 Table S2. Primer pairs for measuring mRNA levels of the targeted genes in U251

 cells



Figure S2 <sup>13</sup>C NMR spectrum of perforalactone D (1) in CDCl<sub>3</sub>.



Figure S3 <sup>1</sup>H-<sup>1</sup>H COSY spectrum of perforalactone D (1) in CDCl<sub>3</sub>.



Figure S4 HSQC spectrum of perforalactones D (1) in CDCl<sub>3</sub>.



Figure S5 HMBC spectrum of perforalactone D (1) in CDCl<sub>3</sub>.



Figure S6 ROESY spectrum of perforalactone D (1) in CDCl<sub>3</sub>.

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Figure S7 HR-ESI-MS spectrum of perforalactone D (1)



Figure S8 IR (KBr disk) spectrum of perforalactone D (1)



Figure S9 ECD spectrum of perforalactone D (1) in methanol.



Figure S10<sup>1</sup>H NMR spectrum of perforalactone E (2) in CDCl<sub>3</sub>.



Figure S11 <sup>13</sup>C NMR spectrum of perforalactone E (2) in CDCl<sub>3</sub>.



Figure S12 <sup>1</sup>H-<sup>1</sup>H COSY spectrum of perforalactone E (2) in CDCl<sub>3</sub>.



Figure S13 HSQC spectrum of perforalactone E (2) in CDCl<sub>3</sub>.



Figure S14 HMBC spectrum of perforalactone E (2) in CDCl<sub>3</sub>.



Figure S15 ROESY spectrum of perforalactone E (2) in CDCl<sub>3</sub>.

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Printed at: 10:47 AM on: 4/16/2020





Figure S17 IR (KBr disk) spectrum of perforalactone E (2)



Figure S18 ECD spectrum of perforal actone E(2) in methanol.